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REPUBLIEK VAN SUID-AFRIKA

Certificate

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REC'D 22 MAR 2004

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the documents annexed hereto are true copies of:

Application forms P.1 and P.3, provisional specification and  
drawings of South African Patent Application No. 2002/6736  
as originally filed in the Republic of South Africa on 22 August  
2002 in the name of **SASOL TECHNOLOGY (PROPRIETARY)**  
**LIMITED** for an invention entitled: "HYDROGENATION OF  
OLEFINIC FEEDSTOCKS".

Geteken te  
Signed at

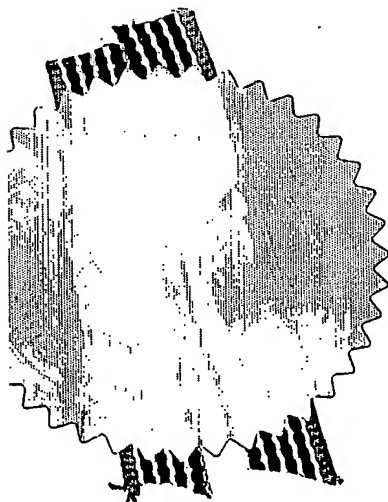
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THE GRANT OF A PATENT IS HEREBY REQUESTED BY THE UNDERMENTIONED APPLICANT  
ON THE BASIS OF THE PRESENT APPLICATION FILED IN DUPLICATE

21 01 PATENT APPLICATION 2002/6736

A&A REF 44/53667 GSK

71 FULL NAME(S) OF APPLICANT(S)

SASOL TECHNOLOGY (PROPRIETARY) LIMITED

ADDRESS(ES) OF APPLICANT(S)

1 Sturdee Avenue, Rosebank, Johannesburg, Republic of South Africa

54 TITLE OF INVENTION

"HYDROGENATION OF OLEFINIC FEEDSTOCKS"

Only the items marked with an "X" in the blocks below are applicable.

THE APPLICANT CLAIMS PRIORITY AS SET OUT ON THE ACCOMPANYING FORM P.2. The earliest priority claimed is

Country:

No:

Date:

THE APPLICATION IS FOR A PATENT OF ADDITION TO PATENT APPLICATION NO

21 01

THIS APPLICATION IS A FRESH APPLICATION IN TERMS OF SECTION 37 AND BASED ON

APPLICATION NO

21 01

THIS APPLICATION IS ACCOMPANIED BY:

- ☒ A single copy of a provisional specification of 27 pages
- ☒ Drawings of 1 sheets
- ☐ Publication particulars and abstract (Form P.8 in duplicate) (for complete only)
- ☐ A copy of Figure of the drawings (if any) for the abstract (for complete only)
- ☐ An assignment of invention
- ☐ Certified priority document(s). (State quantity)
- ☐ Translation of the priority document(s)
- ☐ An assignment of priority rights
- ☐ A copy of Form P.2 and the specification of RSA Patent Application No
- ☐ Form P.2 in duplicate
- ☐ A declaration and power of attorney on Form P.3
- ☐ Request for ante-dating on Form P.4
- ☐ Request for classification on Form P.9
- ☐ Request for delay of acceptance on Form P.4
- ☐ Extra copy of informal drawings (for complete only)

21 01

74 ADDRESS FOR SERVICE: Adams & Adams, Pretoria

Dated this 22 day of August 2002

ADAMS & ADAMS  
APPLICANTS PATENT ATTORNEYS

The duplicate will be returned to the applicant's address for service as  
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OFFICIAL DATE STAMP	
REGISTRATION OF PATENTS, DESIGNS, TRADE MARKS AND COPYRIGHT	
2002 -08- 22	
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PATENT APPLICATION NO		
21	01	2002/6736

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LODGING DATE	
22	22 AUGUST 2002

FULL NAME(S) OF APPLICANT(S)	
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71	SASOL TECHNOLOGY (PROPRIETARY) LIMITED
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FULL NAME(S) OF INVENTOR(S)	
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72	STARK, NICOLAUS LADISLAUS SWART, JOSIAS SEVAAS DE KOCK
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EARLIEST PRIORITY CLAIMED	COUNTRY	NUMBER	DATE
33	---	31	---

NOTE: The country must be indicated by its International Abbreviation - see schedule 4 of the Regulations

TITLE OF INVENTION	
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54	"HYDROGENATION OF OLIFINIC FEEDSTOCKS"
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\* I/We Morné Barradas

hereby declare that :-

1. ~~I/we am/are the applicant(s) mentioned above;~~

\*\* 2. I/we have been authorized by the applicant(s) to make this declaration and have knowledge of the facts herein stated in the capacity of Authorized Signatory of the applicant(s);

\*\*\* 3. the inventor(s) of the abovementioned invention is/are the person(s) named above and the applicant(s) has/have acquired the right to apply by virtue of an assignment from the inventor(s);

4. to the best of my/our knowledge and belief, if a patent is granted on the application, there will be no lawful ground for the revocation of the patent;

\*\*\* 5. ~~this is a convention application and the earliest application from which priority is claimed as set out above is the first application in a convention country in respect of the invention claimed in any of the claims; and~~

6. the partners and qualified staff of the firm of ADAMS & ADAMS, patent attorneys, are authorised, jointly and severally, with powers of substitution and revocation, to represent the applicant(s) in this application and to be the address for service of the applicant(s) while the application is pending and after a patent has been granted on the application.

SIGNED THIS 31 DAY OF October

2002

Company Name: SASOL TECHNOLOGY (PROPRIETARY) LIMITED  
Full Names: Morné Barradas  
Capacity: Authorized Signatory

(no legalization necessary)

\* In the case of application in the name of a company, partnership or firm, give full names of signatory/signatories, delete paragraph 1, and enter capacity of each signatory in paragraph 2.  
\*\* If the applicant is a natural person, delete paragraph 2.  
\*\*\* If the right to apply is not by virtue of an assignment from the inventor(s), delete "an assignment from the inventor(s)" and give details of acquisition of right.  
For non-convention applications, delete paragraph 5.

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PATENT ATTORNEYS  
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FORM P6

REPUBLIC OF SOUTH AFRICA  
Patents Act, 1978

**PROVISIONAL SPECIFICATION**

(Section 30 (1) - Regulation 27)

21	01	OFFICIAL APPLICATION NO
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22	LODGING DATE
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22 August 2002

71	FULL NAME(S) OF APPLICANT(S)
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SASOL TECHNOLOGY (PROPRIETARY) LIMITED

72	FULL NAME(S) OF INVENTOR(S)
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STARK, NICOLAUS LADISLAUS  
SWART, JOSIAS SEVAAS DE KOCK

54	TITLE OF INVENTION
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"HYDROGENATION OF OLEFINIC FEEDSTOCKS"

5

THIS INVENTION relates to the hydrogenation of olefinic feedstocks. In particular, it relates to a process for hydrogenating an olefinic feedstock containing a plurality of different unsaturated olefinic hydrocarbon compounds.

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According to the invention, there is provided a process for hydrogenating an olefin-containing feedstock containing a plurality of different unsaturated olefinic hydrocarbon compounds, the process including

subjecting the olefinic feedstock to bulk hydrogenation by means of catalytic distillation in a catalytic distillation zone containing a hydrogenation catalyst, and in the presence of hydrogen, thereby to hydrogenate unsaturated olefinic hydrocarbon compounds present in the feedstock into their corresponding saturated compounds; and

withdrawing the saturated compounds from the catalytic distillation zone.

20

The catalytic distillation in the catalytic distillation zone involves effecting hydrogenation reactions, under the influence of the hydrogenation catalyst, simultaneously with, or in combination with, distillation in the same zone. In other words, hydrogenation and separation by means of distillation, are effected simultaneously in a single zone.

25

By 'bulk hydrogenation' is meant that a number of different unsaturated olefinic hydrocarbon compounds, that are present in the olefinic feedstock, are hydrogenated. In one embodiment of the invention, substantially all the unsaturated olefinic hydrocarbon compounds in the feedstock may be hydrogenated. The process may then include feeding

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the saturated compounds into a separation stage, and separating lighter saturated compounds or paraffins from heavier saturated compounds or paraffins.

5 In another embodiment of the invention, however, the number of different unsaturated olefinic hydrocarbon compounds that are hydrogenated may be less than the total number of different unsaturated olefinic hydrocarbon compounds present in the feedstock. The process may then include withdrawing at least one unreacted unsaturated olefinic  
10 hydrocarbon compound from the catalytic distillation zone.

The invention is thus characterized thereby that, by means of the bulk hydrogenation, a number of different hydrocarbon compounds are hydrogenated, rather than only a single unsaturated hydrocarbon  
15 compound or a single category of unsaturated hydrocarbon compounds, such as dienes and/or acetylenic compounds, being hydrogenated.

The feedstock may comprise from 60% by mass to 100% by mass unsaturated olefinic hydrocarbon compounds, typically from 80% to  
20 100% by mass of such compounds. When the feedstock comprises less than 100% by mass of unsaturated olefinic hydrocarbon compounds, the balance of the feedstock may typically be made up of branched and normal paraffins such as octane and/or 2-methyl heptane; oxygenates such as alcohols; aromatics such as benzene; and saturated and  
25 unsaturated cyclic compounds other than aromatics, such as cyclohexene. Any aromatic compounds present in the feedstock will be hydrogenated to the corresponding saturated cyclic compounds; however, any oxygenated compounds present in the feedstock will not normally be hydrogenated during the bulk hydrogenation of the feedstock.

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In one embodiment of the invention, the feedstock may be a C<sub>7</sub>-C<sub>13</sub> naphtha feedstock, ie it may contain a range of different unsaturated

olefinic hydrocarbon compounds having from 7 to 13 carbon atoms. However, in another embodiment of the invention, the feedstock may comprise oligomers obtained from C<sub>3</sub>-C<sub>7</sub> unsaturated olefinic hydrocarbons, ie it may contain a range of different unsaturated olefinic oligomers.

The feedstock may be Fischer-Tropsch derived, ie it may be obtained from the so-called Fischer-Tropsch process. In other words, it may be obtained by reacting a synthesis gas comprising carbon monoxide and hydrogen in the presence of a suitable Fischer-Tropsch catalyst, normally a cobalt, iron, or cobalt/iron Fischer-Tropsch catalyst, at elevated temperature in a suitable reactor, which is normally a fixed or slurry bed reactor, thereby to obtain a range of products, including a range of olefinic or unsaturated hydrocarbon compounds suitable for use as the feedstock in this invention. The products from the Fischer-Tropsch process must then usually be worked up to obtain the olefinic feedstock. Thus, as hereinbefore described, the feedstock will then normally contain, in addition to the unsaturated olefinic hydrocarbon compounds, also branched and normal paraffins such as octane and/or 2-methyl heptane; oxygenates such as alcohols; aromatics such as benzene; and saturated and unsaturated cyclic compounds other than aromatics, such as cyclohexene.

From 30% to about 100% of the unsaturated olefinic hydrocarbon compounds may be hydrogenated in the catalytic distillation zone.

While a single unsaturated olefinic hydrocarbon compound may, at least in principal, remain unhydrogenated or unreacted in the catalytic distillation zone, two or more different unsaturated olefinic hydrocarbon compounds will normally remain unhydrogenated or unreacted. These unreacted or unhydrogenated compounds are usually either the lightest compounds in the feedstock or the heaviest compounds in the feedstock,

with the process of the invention thus resulting in these compounds being separated, in the catalytic distillation zone, from the hydrogenated compounds.

- 5 The catalytic distillation zone is typically provided by a column. The catalyst may be in particulate form, and may be provided in the form of a packed bed. The feedstock and the hydrogen will then naturally be fed continuously into the column, with the product being withdrawn continuously from the column as a product stream. The feedstock and
- 10 hydrogen may enter the column at about the same level, or at different levels. Preferably, however, the hydrogen may enter the column at a level below the level at which the feedstock enters the column. Suitable distillation media, eg random packing, structured packing, trays or any other distillation apparatus or equipment, are provided in the column
- 15 below and/or above the catalyst bed.

- The particulate hydrogenation catalyst in the packed bed permits good contact between the unsaturated olefinic hydrocarbon compounds and the hydrogen, while also providing the required separation between gas and
- 20 liquid phases. The hydrogenation catalyst may, in particular, be a heterogeneous catalyst. It typically has a particle size of 0.79mm to 6.35mm, and may be of any desired shape, eg spherical, elongate or the like. Typically, such catalysts contain one or more metals such as nickel, copper, cobalt, chromium, zinc, iron and the platinum group metals, ie
- 25 platinum, palladium, rhodium and ruthenium, as their active component.

- While the column can, at least in principle, operate at an elevated pressure of up to 1500kPa(g), it is envisaged that it will normally operate at about atmospheric pressure or at only slightly above atmospheric
- 30 pressure, which is an advantage of the process according to the invention. Thus, the operating pressure in the column may typically be in the range of about 50kPa(g) to about 200kPa(g).



The temperature in the column will be dependent on; amongst others, the feedstock composition, the column pressure and the unsaturated olefinic hydrocarbon compounds which are not to be hydrogenated, ie which are to be separated from the hydrogenated compounds. Thus, when the feedstock is the C<sub>7</sub> – C<sub>13</sub> naphtha feedstock, and the column operating pressure is 100kPa(g)-200kPa(g), the catalyst bed temperature may be about 120°C-140°C, with the product stream then being removed as a bottoms stream and the unreacted unsaturated olefinic hydrocarbon compounds being lighter compounds which are removed as an overheads stream. When the feedstock is the C<sub>7</sub> – C<sub>13</sub> naphtha feedstock, and the column operating pressure is about 100kPa(g), the catalyst bed temperature may be about 150°C, with the product stream being removed as an overheads stream and the unreacted unsaturated olefinic hydrocarbon compounds being heavier compounds which are removed as a bottoms stream. When the feedstock comprises unsaturated olefinic oligomers derived from C<sub>3</sub> – C<sub>7</sub> olefins, and the column operating pressure is about 50kPa(g) – 200kPa(g), the catalyst bed temperature may be about 160°C-200°C, with the product stream being removed as an overheads stream and the unreacted unsaturated hydrocarbon compounds being heavier compounds which are removed as a bottoms stream.

The invention will now be described by way of example with reference to the accompanying drawings.

In the drawings,

FIGURE 1 shows a simplified flow diagram of a process for hydrogenating an olefinic feedstock, according to a first embodiment of the invention; and

FIGURE 2 shows a simplified flow diagram of a process for hydrogenating an olefinic feedstock, according to a second embodiment of the invention.

Referring to Figure 1, reference numeral 10 generally indicates a process for hydrogenating an olefinic feedstock, according to a first embodiment of the invention.

5

The process 10 includes a catalytic distillation column 12 containing a packed bed 14 of a particulate hydrogenation catalyst as well as a plurality of distillation plates 16 in a distillation region 20 above the packed catalyst bed 14.

10

A feedstock feed line 20 leads into the distillation region 18, while a hydrogen feed line 22 leads into the column 12 below the packed catalyst bed 14.

15

A bottoms withdrawal line 24 leads from the bottom of the column 12. It splits into a reboil line 26 and a product withdrawal line 28. The line 26 is fitted with a reboiler 30, and returns to the bottom portion of the column 12.

20

An overheads line 32 leads from the top of the column 12 to a condenser 34 and from there to a reflux drum 36. A liquids line 38 leads from the reflux drum 36, and splits into a return line 40 to the top of the column 12 and an overheads product withdrawal line 42. A hydrogen withdrawal line 44 leads from the reflux drum 36 to a hydrogen recycle compressor

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46, with a line 48 leading from the compressor 46 to the hydrogen feed line 22. A hydrogen make-up line 50 also leads into the hydrogen feed line 22.

30

In use, an olefinic feedstock, such as a Fischer-Tropsch derived naphtha feedstock, is fed into the distillation region 18 along the feed line 20, while hydrogen is simultaneously fed into the bottom of the column along the line 22. The column 12 is maintained at slightly above atmospheric

pressure, typically at 100kPa(g)-200kPa(g), with the catalyst bed temperature typically being controlled at 120°C-165°C. Heavier unsaturated olefinic hydrocarbon compounds in the Fischer-Tropsch naphtha feedstock are hydrogenated to paraffins, with these paraffins being withdrawn along the line 24 as a bottoms stream or product. Some of the bottoms product is reboiled via the line 26 and the reboiler 30, with the remainder thereof being withdrawn along the line 28. Lighter unreacted or unhydrogenated unsaturated olefinic hydrocarbon products are withdrawn through the line 32 and condensed in the condenser 34 before passing to the reflux drum 36. A liquid component thereof is withdrawn along the line 38 with a portion thereof being refluxed along the line 40 to the top of the column 12 while the remainder is withdrawn as an overheads product or stream along the line 42. Hydrogen is recycled as feed to the column by means of the line 44 leading from the reflux drum 36, the compressor 46 and the line 48.

By means of the process 10, bulk hydrogenation of a Fischer-Tropsch derived naphtha feedstock can thus be effected. In this bulk hydrogenation, heavier unsaturated olefinic hydrocarbon compounds are hydrogenated to paraffins which are withdrawn along the line 28 as the bottoms product. Unwanted lighter unsaturated olefinic hydrocarbon compounds are withdrawn along the line 42 as the overheads product.

It will be appreciated that other feedstocks can be treated in a process having the same configuration as in Figure 1.

Thus, in another version of this embodiment of the invention, the process 10 can be used for bulk hydrogenation of olefinic feedstock comprising unsaturated olefinic oligomers (ie polymers made up of 2, 3 or 4 monomer units) derived from  $C_3 - C_7$  olefins. The unsaturated oligomers are hydrogenated to paraffins, with the paraffins being withdrawn as the bottoms product, and with unwanted light

unhydrogenated or unsaturated olefinic oligomers and bottoms being removed as the overheads product.

In the process 10, the degree of hydrogenation is determined by the supply of hydrogen along the line 22, and the operating conditions of the column 12. Hydrogenation does not necessarily have to be complete. The hydrogen recycle compressor 46 ensures adequate hydrogen partial pressures in the catalyst bed 14.

The process 10 is exemplified in Examples 1 and 2 hereunder. In Examples 1 and 2, as well as in Examples 3 and 4 also described hereunder, a 10 meter 2 inch (approximately 5cm) diameter catalytic distillation column 12, consisting of four 2.5m sections, was used. In each of Examples 1 to 4, the column was loaded with a commercially available hydrogenation catalyst as specified. The particulate catalyst was packed in pockets made from woven stainless steel mesh wrapped in demister wire. The column had feed points on an upper flange of all the 2.5m sections, to allow for process optimization. The hydrogenated compounds could be removed as either an overheads product stream or a bottoms product stream.

Generally, in Examples 1 and 2 hereunder, the process configuration was as indicated in Figure 1 except that the hydrogen recycle lines 44, 48 and the hydrogen compressor 46 were omitted. Instead, a hydrogen purge line led from the reflux drum 36. No distillation plates were provided either above or below the catalyst bed and the catalyst packing thus also fulfilled the roll of distillation plates.

Generally, in Examples 1 and 2, a C<sub>7</sub> – C<sub>13</sub> Fischer-Tropsch derived naphtha feedstock, with an olefin content of about 84 mass %, was fed above the catalyst bed 14, at a rate of 0.5-1kg/h. Hydrogen was fed into the bottom of the column 12 along the line 22, ie below the catalyst bed

14, at a rate of 1.0-2.5m<sup>3</sup>n/hour. The column pressure was varied between 100-200kPa(g) which resulted in catalyst bed temperatures ranging from 120°C-140°C. The conversion of the olefins in the feedstock was about 60-80%. The hydrogenated compounds were removed as a bottoms stream.

It was found that similar results to those obtained in Examples 1 and 2 could also be achieved using a larger scale catalytic distillation column 12 also having a length of 10m but having a 4 inch (about 100mm) diameter, and using the same commercially available hydrogenation catalyst.

#### EXAMPLE 1

The 10 meter 2 inch column was loaded with a commercially available hydrogenation catalyst obtainable from Kata Leuna GmbH Catalysts of Am Haupttor, Geb 8322, D-06237 Leuna, Germany under the designation Leuna-Catalyst 6564TL 1.2. A C<sub>7</sub> – C<sub>13</sub> Fischer-Tropsch derived naphtha feedstock with an olefin content of between 42 and 72 mass % was fed above the catalyst bed at a rate of 1 kg/hr. The feedstock composition is given in Table 1.1.

Table 1.1 Hydrocarbon Feedstock composition and Feed Characterisation.

Feed component	Mass %
C5	0.23
C6	2.18
C7	17.78
C8	27.39
C9	23.78
C10	17.49
C11	9.46
>C11	1.69
Total	100
Feed Characterisation	
Bromine no. (g Br/100 g)	81.6
Acid no. (mg KOH/g)	15.4
Carbonyls (% MEK)	5.9
Alcohols (% C7)	6.5
Esters (mg KOH/g)	2.3

- 5 Hydrogen was fed below the catalyst bed at a rate of 89g/hr. The column pressure was 100kPa(g) which resulted in a catalyst bed temperature of 117°C. The reboiler temperature was 164°C. The hydrogenated compounds were removed as the bottoms stream. 798g of bottoms product and 200g of overheads product were collected per hour. The conversion of the olefins in the feedstock was 57%. The analyses of the overheads and bottoms products are given in Table 1.2 below.

Table 1.2 Product analyses

Overheads	
Bromine no. (g Br/100 g)	15.77
Acid no. (mg KOH/g)	18.90
Carbonyls (% MEK)	6.30
Alcohols (% C7)	10.50
Esters (mg KOH/g)	2.20
Bottoms (Hydrogenated product)	
Bromine no. (g Br/100 g)	40.25
Acid no. (mg KOH/g)	5.90
Carbonyls (% MEK)	4.50
Alcohols (% C7)	3.10
Esters (mg KOH/g)	10.30

5 EXAMPLE 2

The 10 meter 2 inch column was loaded with the same commercially available hydrogenation catalyst as used in Example 1. The same Fischer-Tropsch derived feedstock as used in Example 1, was fed above the catalyst bed at a rate of 748g/hr.

10

Hydrogen was fed below the catalyst bed at a rate of 224g/hr. The column pressure was 212kPa(g) which resulted in a catalyst bed temperature of 140°C. The reboiler temperature was 197°C. The hydrogenated products were removed as the bottoms stream. 544g of bottoms product and 216g of overheads product were collected per hour. The conversion of the olefins in the feed stream was 78%. The analyses of the overheads and bottoms products are given in Table 2.1 below.

15

Table 2.1 Product analyses

Overheads	
Bromine no. (g Br/100 g)	10.25
Acid no. (mg KOH/g)	19.40
Carbonyls (% MEK)	3.60
Alcohols (% C7)	14.80
Esters (mg KOH/g)	2.40
Bottoms (Hydrogenated product)	
Bromine no. (g Br/100 g)	20.82
Acid no. (mg KOH/g)	1.50
Carbonyls (% MEK)	2.70
Alcohols (% C7)	4.10
Esters (mg KOH/g)	18.30

5 Referring to Figure 2, reference numeral 100 generally indicates a process for hydrogenating an olefinic feedstock, according to a second embodiment of the invention.

10 In the process 100, components which are the same or similar to those of the process 10 of Figure 1, are indicated with the same reference numerals.

15 The catalytic distillation column 12 of the process 100 is similar to that of the process 10, except that its distillation region 18 is provided below the packed catalyst bed 14. The feedstock feed line 20 still leads into the distillation zone 18, and it is thus also located below the packed catalyst bed 14.



In the process 100, bulk hydrogenation of an olefinic feedstock, such as a Fischer-Tropsch derived naphtha feedstock can be effected, with unsaturated hydrocarbon compounds in the feedstock being converted to paraffins. The hydrogenated compounds, ie the paraffins, are withdrawn along the line 42 as an overheads product, with unwanted heavier unsaturated hydrocarbon compounds, ie feed oligomers, being removed along the line 28 as a bottoms product.

It will be appreciated that other feedstocks can be treated in a process having the same configuration as in Figure 2.

Thus, in another version of this embodiment of the invention, hydrogenation of a mixture of oligomers derived from C<sub>3</sub> - C<sub>7</sub> unsaturated or olefinic hydrocarbon compounds, can be hydrogenated in the process 100. The unsaturated oligomers are hydrogenated to paraffins. The paraffins were withdrawn as an overheads product, with unwanted unsaturated heavier components, in the form of heavier olefins and/or oligomers, were removed as a bottoms product. As before, the degree of hydrogenation is determined by the supply of hydrogen and the operating conditions in the catalytic distillation column 12, and the hydrogenation does not necessarily have to be complete. The hydrogen recycle compressor 46 ensures adequate hydrogen partial pressures in the packed bed 14 of the catalytic distillation column 12.

As in Figure 1, the degree of hydrogenation is determined by the supply of hydrogen and the operating conditions in the column 14; the hydrogenation does not necessarily have to be complete; and the hydrogen recycle compressor 46 ensures adequate hydrogen partial pressures in the catalyst bed 14.

In Examples 3 to 9 hereunder, the process 100 was used save that, instead of the hydrogen recycle along the line 44, the compressor 46 and the line 48, a hydrogen purge from the reflux drum 36 was used.

- 5 Generally, in Examples 3 and 4, a C<sub>7</sub> – C<sub>13</sub> Fischer-Tropsch naphtha feedstock with an olefin content of 84 mass % was fed into the catalytic distillation column 12 below the catalyst bed 14, at a rate of 2kg/h. Hydrogen was fed into the column 14 at a rate of 2m<sup>3</sup>n/hour, below the catalyst bed 14. The column pressure was maintained at 100kPa(g) which resulted in catalyst bed temperatures of about 150°C. The  
10 conversion of olefins in the feedstock was 80%-85%.

### EXAMPLE 3

- The 10 meter 2 inch column was loaded with the same commercially  
15 available hydrogenation catalyst as used in Example 1. A C<sub>7</sub> – C<sub>13</sub> Fischer-Tropsch naphtha feedstock with an olefin content of between 42 and 72 mass % was fed below the catalyst bed at a rate of 2 kg/hr. The feed composition is given in Table 3.1.

Table 3.1 Hydrocarbon Feedstock composition and Feed Characterisation.

Feed component	Mass %
C8	1.00
C9	49.43
C10	33.13
C11	15.73
C12	0.71
>C12	0.00
Total	100
	-
Feed Characterisation	
Bromine no. (g Br/100 g)	82.5
Acid no. (mg KOH/g)	0.2
Carbonyls (% MEK)	5.9
Alcohols (% C7)	6.2
Esters (mg KOH/g)	2.6

- 5 Hydrogen was fed below the catalyst bed at a rate of 179g/hr. The column pressure was 100kPa(g) which resulted in catalyst bed temperature of 143°C. The reboiler temperature was 219°C. The reflux ratio was maintained at 2. The hydrogenated compounds were removed as the overheads stream. 486g of bottoms product and
- 10 1.497kg of overheads product were collected per hour. Excess hydrogen was flared. The conversion of the olefins in the feedstock was 83%. The analyses of the overheads and bottoms products are given in Table 3.2 below.

Table 3.2 Product analyses

Overheads (Hydrogenated product)	
Bromine no. (g Br/100 g)	3.25
Acid no. (mg KOH/g)	0.02
Carbonyls (% MEK)	5.05
Alcohols (% C7)	5.70
Esters (mg KOH/g)	2.50
Bottoms	
Bromine no. (g Br/100 g)	47.18
Acid no. (mg KOH/g)	0.09
Carbonyls (% MEK)	3.85
Alcohols (% C7)	1.85
Esters (mg KOH/g)	4.50

5 EXAMPLE 4

The 10 meter 2 inch column was loaded with the same commercially available hydrogenation catalyst as used in Example 1. The same Fischer-Tropsch feedstock as used in Example 3 was fed below the catalyst bed at a rate of 2kg/hr.

10

Hydrogen was fed below the catalyst bed at a rate of 179g/hr. The column pressure was 100kPa(g) which resulted in catalyst bed temperature of 148°C. The reboiler temperature was 236°C. The reflux ratio was maintained at 2. The hydrogenated compounds were removed as the overheads stream. 102g of bottoms product and 1.897kg of overheads product were collected per hour. Excess hydrogen was flared. The conversion of the olefins in the feed stream was 75%. The analyses of the overhead and bottom products are given in Table 4.1 below.

15

Table 4.1 Product analyses

Overheads (Hydrogenated product)	
Bromine no. (g Br/100 g)	20.14
Acid no. (mg KOH/g)	0.11
Carbonyls (% MEK)	5.60
Alcohols (% C7)	5.70
Esters (mg KOH/g)	2.25
Bottoms	
Bromine no. (g Br/100 g)	21.83
Acid no. (mg KOH/g)	0.13
Carbonyls (% MEK)	4.95
Alcohols (% C7)	0.40
Esters (mg KOH/g)	5.45

5

In Examples 5 to 9, a 10 meter 4 inch (approximately 100mm) diameter catalytic distillation column 12, consisting of 10 1m sections, was used. In each of the Examples, it was loaded with a packed bed 14 of commercially available hydrogenation catalyst as specified. The catalyst was packed in proprietary catalyst packing obtainable from Catalytic Distillation Technologies of 10100 Bay Area Boulevard, Pasadena, Texas 77507, United States of America, and described in US 5942456. The column had feed points on an upper flange of all the 1m sections, to allow for process optimization. The hydrogenated compounds were removed as a bottoms stream.

EXAMPLE 5

The 10 meter 4 inch column was loaded with a commercially available hydrogenation catalyst obtainable from Kata Leuna Catalysts of Am

Haupttor, Geb 8322, D-06237 Leuna, Germany under the designation Leuna-Catalyst 7762K. A C<sub>7</sub> – C<sub>13</sub> Fischer-Tropsch derived naphtha feedstock with an olefin content of between 45 and 80 mass % was fed below the catalyst bed at a rate of 14.251 kg/hr. The feed composition is given in Table 5.1.

Table 5.1 Hydrocarbon Feedstock composition and Feed Characterisation.

Feed component	Mass %
C6	0.42
C7	15.73
C8	27.48
C9	24.50
C10	17.68
C11	11.30
C12	2.03
>C12	0.06
Total	100
Feed Characterisation	
Bromine no.(g Br/100 g)	90.0
Acid no. (mg KOH/g)	17.2
Carbonyls (% MEK)	6.4
Alcohols (% C7)	6.3
Esters (mg KOH/g)	4.3

10

Hydrogen was fed below the catalyst bed at a rate of 396g/hr. The column pressure was 102kPa(g) which resulted in catalyst bed temperature of 172°C. The reboiler temperature was 203°C. The reflux ratio was maintained at 6. The hydrogenated compounds were removed as the overhead stream. 2.789kg of bottoms product and 11.463kg of overheads product were collected per hour. Excess hydrogen was flared. The conversion of the olefins in the feedstock was 87%. The analyses of the overheads and bottoms products are given in Table 5.2 below.

20

Table 5.2 Product analyses

Overheads (Hydrogenated product)	
Bromine no.(g Br/100 g)	10.12
Acid no. (mg KOH/g)	16.30
Carbonyls (% MEK)	5.30
Alcohols (% C7)	6.80
Esters (mg KOH/g)	3.70
Bottoms	
Bromine no.(g Br/100 g)	17.55
Acid no. (mg KOH/g)	1.30
Carbonyls (% MEK)	3.00
Alcohols (% C7)	0.60
Esters (mg KOH/g)	32.00

5

EXAMPLE 6

The 10 meter 4 inch column was loaded with the same commercially available hydrogenation catalyst as was used in Example 5. The same Fischer-Tropsch derived naphtha feedstock as used in Example 5 was fed below the catalyst bed at a rate of 18.016kg/hr.

Hydrogen was fed below the catalyst bed at a rate of 434g/hr. The column pressure was 300kPa(g) which resulted in catalyst bed temperature of 208°C. The reboiler temperature was 244°C. The reflux ratio was maintained at 4. The hydrogenated compounds, ie paraffins, were removed as the overhead stream. 2.727kg of bottoms and 15.648kg of overheads were collected per hour. Excess hydrogen was flared. The conversion of the olefins in the feed stream

was 95%. The analyses of the overhead and bottom products are given in Table 6.1 below.

Table 6.1 Product analyses

Overheads (Hydrogenated product)	
Bromine no. (g Br/100 g)	2.18
Acid no. (mg KOH/g)	15.25
Carbonyls (% MEK)	4.95
Alcohols (% C7)	6.45
Esters (mg KOH/g)	2.55
Bottoms	
Bromine no. (g Br/100 g)	14.76
Acid no. (mg KOH/g)	0.47
Carbonyls (% MEK)	2.70
Alcohols (% C7)	0.58
Esters (mg KOH/g)	40.45

Generally, in Examples 7 to 9, an olefinic feedstock comprising an unsaturated oligomer mixture was fed below the catalyst bed 14 at a rate between 5- 15kg/h. Hydrogen was fed at a rate of 1-9m<sup>3</sup>n/h normal per hour below the catalyst bed. The column pressure was varied between 50-200kPa(g), which resulted in catalyst bed temperatures ranging from 160°C-200°C. The conversion of the olefins in the feed stream was 60-99%.

#### EXAMPLE 7

The 10 meter 4 inch column was loaded with the same commercially available hydrogenation catalyst as used in Example 5. An oligomer mixture with an olefin content of between 45 and 80 mass % was



fed, as an olefinic feedstock, below the catalyst bed at a rate of 15.02 kg/hr. The feed composition is given in Table 7.1.

Table 7.1 Hydrocarbon Feedstock composition and Feed Characterisation.

Feed component	Mass %
Propane	0.01
Isobutane	0.27
1-butene + Isobutene	0.54
Butane	5.62
t-2-Butene	3.43
c-2-Butene	2.32
C5 paraffins	1.31
C5 olefins	2.57
C6 paraffins	0.28
C6 olefins	3.83
C7 and heavier hydrocarbons	79.83
C4 and lighter hydrocarbons	6.29
Total	100
Feed Characterisation	
Bromine no.(g Br/100 g)	90.00
Sample 1 RVP	66kPa
Sample 2 RVP	65kPa

Hydrogen was fed below the catalyst bed at a rate 0.79 of kg/hr.

- 10 The column pressure was 163kPa(g) which resulted in catalyst bed temperature of 193°C. The reboiler temperature was 234°C. The reflux flow was maintained at 55kg/hr. The hydrogenated product was removed as the overhead stream. 14.62kg of overheads were collected per hour. Excess hydrogen was flared. The conversion of
- 15 the olefins in the feed stream was 99.9%. The bromine number of the overheads (hydrogenated) product was 0.05.

EXAMPLE 8

The 10 meter 4 inch column was loaded with the same commercially available hydrogenation catalyst as used in Example 5. The same feedstock as was used in Example 7, was fed below the catalyst bed at a rate of 15.00kg/hr.

Hydrogen was fed below the catalyst bed at a rate 0.18 of kg/hr. The column pressure was 133kPa(g) which resulted in catalyst bed temperature of 202°C. The reboiler temperature was 229°C. The reflux flow was maintained at 40kg/hr. The hydrogenated compounds were removed as the overhead stream. 14.80kg of overheads were collected per hour. Excess hydrogen was flared. The conversion of the olefins in the feed stream was 39.0%. The bromine number of the overheads (hydrogenated) product was 54.92.

EXAMPLE 9

The 10 meter 4 inch column was loaded with the same commercially available hydrogenation catalyst as used in Example 5. The same feedstock as was used in Example 7, was fed below the catalyst bed at a rate of 10.02kg/hr.

Hydrogen was fed below the catalyst bed at a rate 0.33 of kg/hr. The column pressure was 52kPa(g) which resulted in catalyst bed temperature of 177°C. The reboiler temperature was 214°C. The reflux flow was maintained at 35kg/hr. The hydrogenated product was removed as the overhead stream. 10.22kg of overheads were collected per hour. Excess hydrogen was flared. The conversion of the olefins in the feed stream was 99.4%. The bromine number of the overhead (hydrogenated) product was 0.56.

EXAMPLES 10-13

Examples 10-13 were performed in identical fashion to Examples 7 to 9, using the same feedstock, catalyst, etc, but having different feedstock feed rates, hydrogen feed rates and other operating parameters. The flow rates, operating parameters, product analyses and results are given in Table 10.1. For completeness, Examples 7 to 9 are included in Table 10.1.

Table 10.1

Period #			7	10	11	8	12	13	9
<b>FLOW RATES</b>									
Feed									
Catpoly	Kg/hr		15.02	14.99	15.01	15.00	10.00	10.00	10.02
Hydrogen	Kg/hr		0.79	0.22	0.29	0.18	0.79	0.79	0.33
Products									
Bottoms	Kg/hr		0.56	*	1.02	0.16	*	0.30	*
Overheads	Kg/hr		14.62	15.20	14.02	14.80	10.14	9.78	10.22
Flare – hydrogen	Kg/hr		0.62	0.12	0.14	0.11	0.68	0.68	0.22
<b>OPERATION</b>									
Column pressure	kPa(g)		163	136	153	133	56	55	52
Catalyst bed temperature	°C		193	201	200	202	159	164	177
Reboiler temperature	°C		234	229	230	229	211	215	214
Reflux flow	kg/hr		55	45	45	40	35	35	35
<b>ANALYSIS</b>									
Products	mass %								
Overheads									
Bromine No	gr Br/100 gr		0.05	35.42	9.45	54.92	2.11	1.15	0.56
<b>RESULTS</b>									
CONVERSION TO PARAFFINS	%								
Total olefins based on Br No			99.9	60.6	89.5	39.0	97.7	98.7	99.4

\* In these Examples, the bottoms production was negligible

The Applicant believes that the process according to the invention has, amongst others, the following advantages:

- 5       - Lower operating pressures can be used in the catalytic distillation column 12 compared to fixed bed/trickle bed hydrogenation technologies, to achieve the same conversion/productivity potential resulting in less capital intensive equipment being required.
- 10       - Hydrogenation is an exothermic reaction so that substantial amounts of heat of reaction are produced. With the process 10, in situ removal of these substantial amounts of heat of reaction can be achieved. Highly liquid recycles or the use of intercoolers is not required, potentially resulting in process simplifications.
- 15       - Due to excellent removal of heat of reaction, ie no hot spots, less fouling of the catalyst due to the formation of oligomers, occurs; this results in an extended catalyst life compared to the same catalyst used in a fixed bed hydrogenation reactor.
- 20       - Acidity of the feed, which can lead to the formation of heavy components/oligomers, has no negative effect on the catalyst activity as the heavy components are continuously washed from the surfaces of the catalyst particles.

25       Additionally, the process according to the invention has general advantages over conventional processes for hydrogenating olefinic feedstocks comprising a hydrogenation reactor followed by a distillation column, such as

- 30       - Equilibrium constraints are overcome, as products are continuously removed from the reaction zone, resulting in increased productivity.

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- Extended catalyst life is expected, due to the removal of products from the catalyst surface as a result of the washing action of the reflux in the catalytic distillation column.

5 - Increased selectivities are expected since high local temperatures, which can lead to by-product formation, are limited.

- Strategic location of feed points into the catalytic distillation column can limit the harmful effect of poisons and/or inhibitors in the feedstock.

10 - The process can handle azeotropic systems.

- The process has the ability to remove large amounts of heat of reaction while maintaining a stable catalyst temperature as the column temperature is set by the column pressure, provided that the column is operated above the minimum required loading.

15 - The general process scheme can be simplified since the two operations of the known processes are now performed in one vessel.

- The heat of reaction is used for separation purposes, resulting in reduced reboiler requirements.

20 DATED THIS 22<sup>ND</sup> DAY OF AUGUST 2002.

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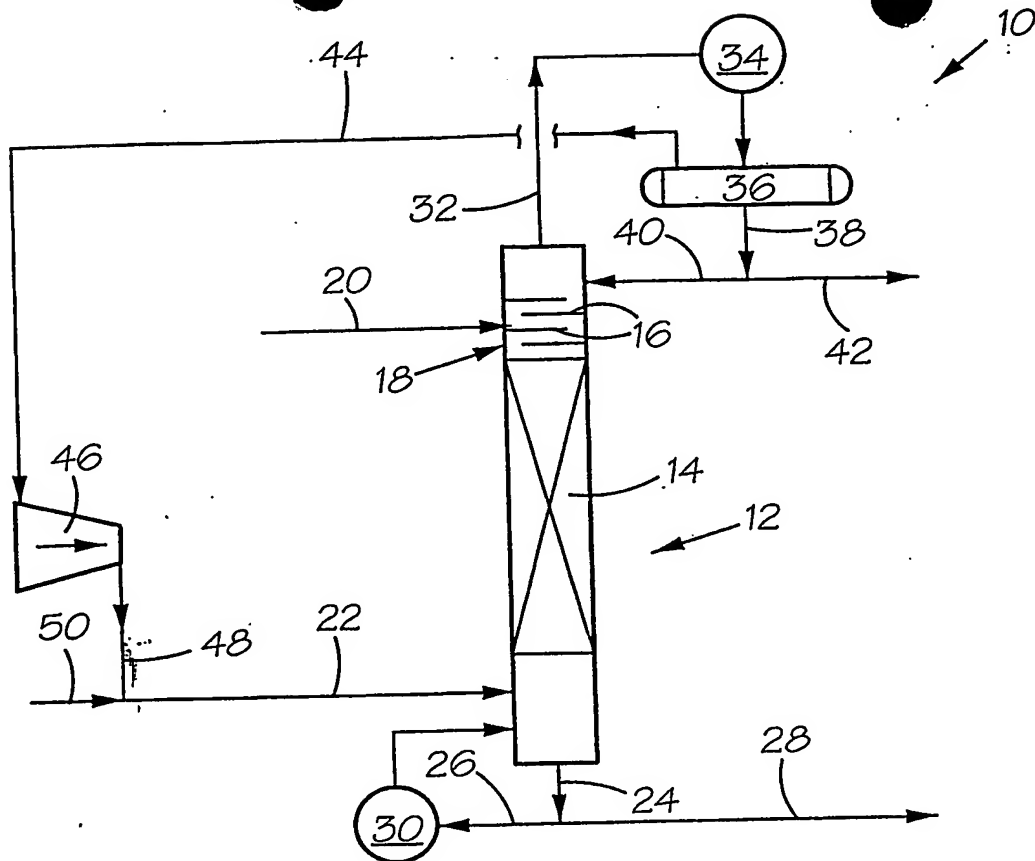


FIG 1

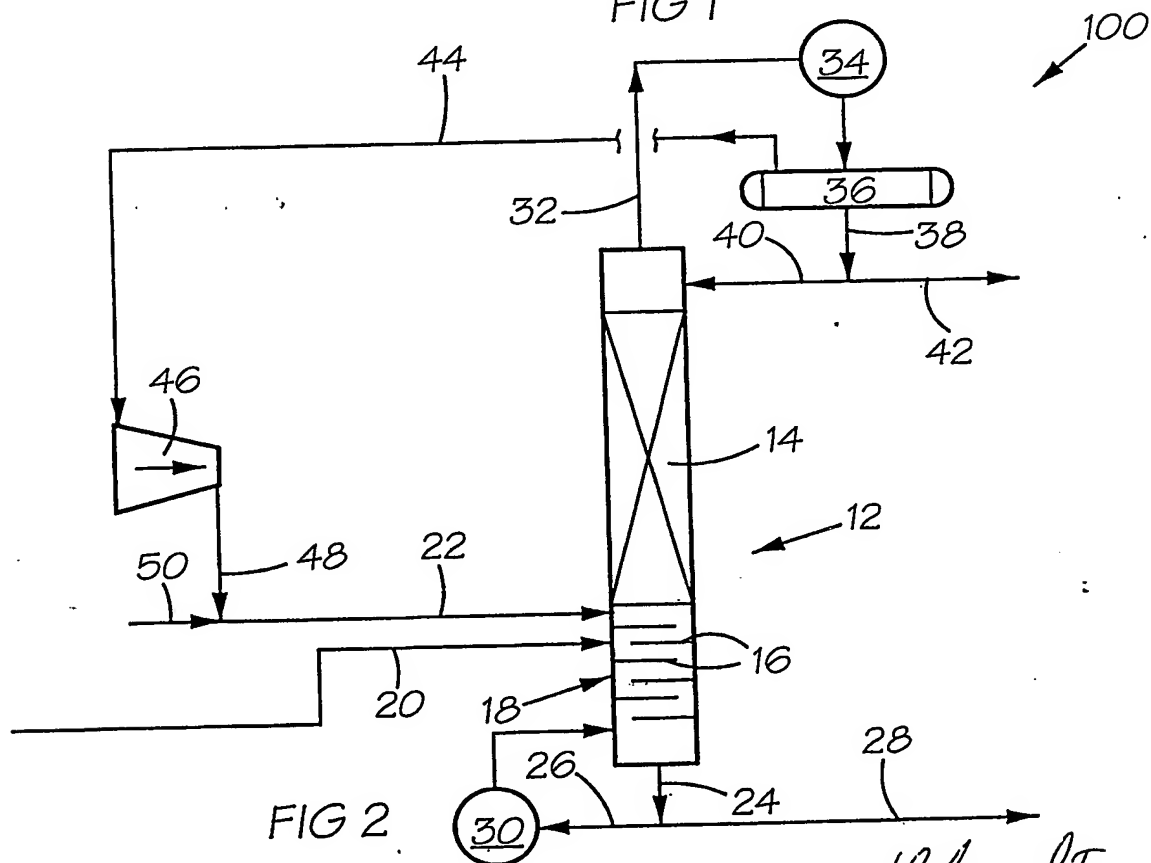


FIG 2

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